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**Quantitative  $^1\text{H}$  qNMR Method for Complex Mixture Analysis:  
Determination of Acetylated Polysaccharides, Glucose,  
Maltodextrin, Isocitrate, Malic Acid in Aloe Vera Leaf Juice.**

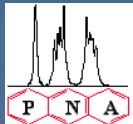
**Also ... Preservatives, Additives, and Degradation Products in  
Aloe Vera Raw Materials and Consumer Products**

**Presented By**

**John Edwards, Ph.D.**

**Process NMR Associates, LLC  
Danbury, Connecticut**

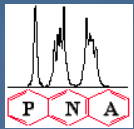
**November 9, 2012  
Aloe Summit, Las Vegas, NV**



### NMR Systems and Experimental Conditions

Acquisition Time	3-8 Seconds
Relaxation (Recycle) Delay	2-6 Seconds
Frequency, MHz	300-500 MHz
Nucleus	$^1\text{H}$
Number of Pulse Accumulations*	16-256
Original FID Points	16384-84000
Zero-filled Points	32768-262144
Pulse sequence	Single pulse
Solvent	D <sub>2</sub> O
Sweep width, ppm	16
Temperature	Ambient (25 °C)
Line Broadening	0.35 Hz
Steady State Pulses	8
Pre-Acquisition Delay	60 seconds

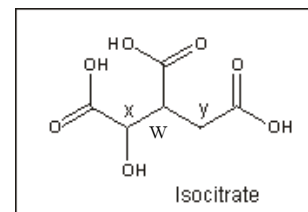




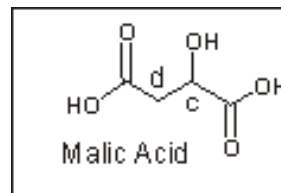
## Advantages of Using NMR in Complex Mixture Analysis

- **Minimal Sample Preparation**
- **Reproducible (site to site, country to country)**
- **Linear and Quantitative Spectral Response**
- **Targeted AND Non-Targeted**
- **Qualitative**
- **Quantitative**
- **Compound Specific**
- **Rapid – 10 Minutes**
- **Affordable**

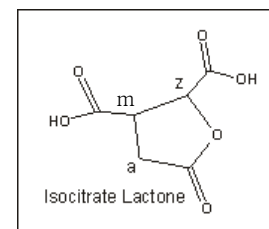
DL-Isocitrate\_Na\_Salt  
DL-Isocitrate TriSodium Salt  
1H NMR in D2O  
JCE-PNA-Merc300



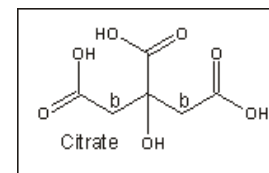
Malic-001-H  
L-Malic Acid  
1H NMR in D2O  
JCE-PNA-MVX300



IsocitrateLactone-001-H  
Isocitrate Lactone Standard (Aldrich)  
1H NMR in D2O  
JCE-PNA-MVX300

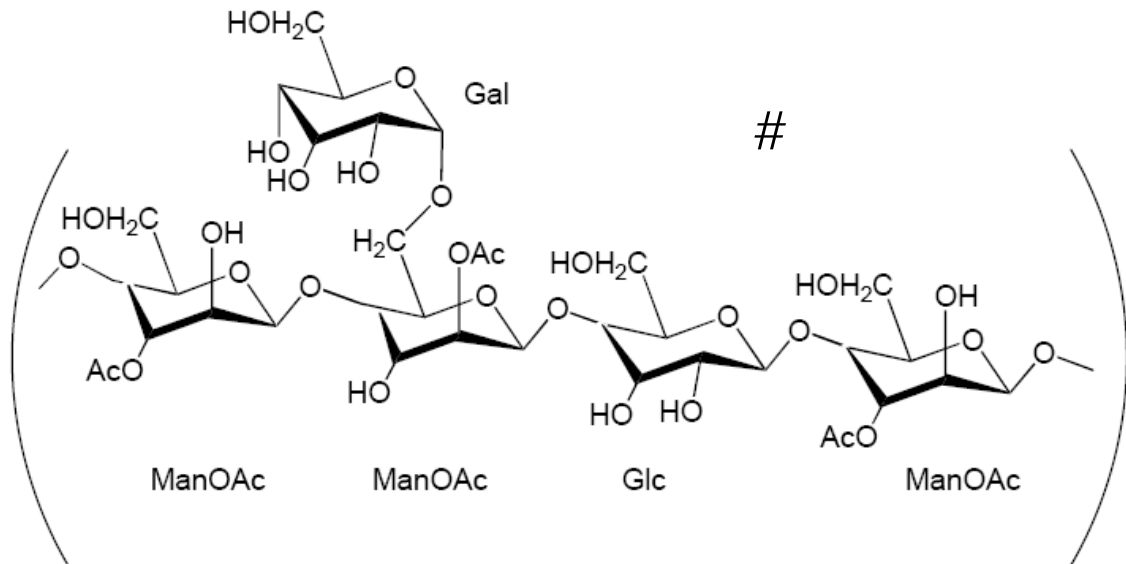


Citric-001-H  
Citric Acid  
1H NMR in D2O  
JCE-PNA-MVX300

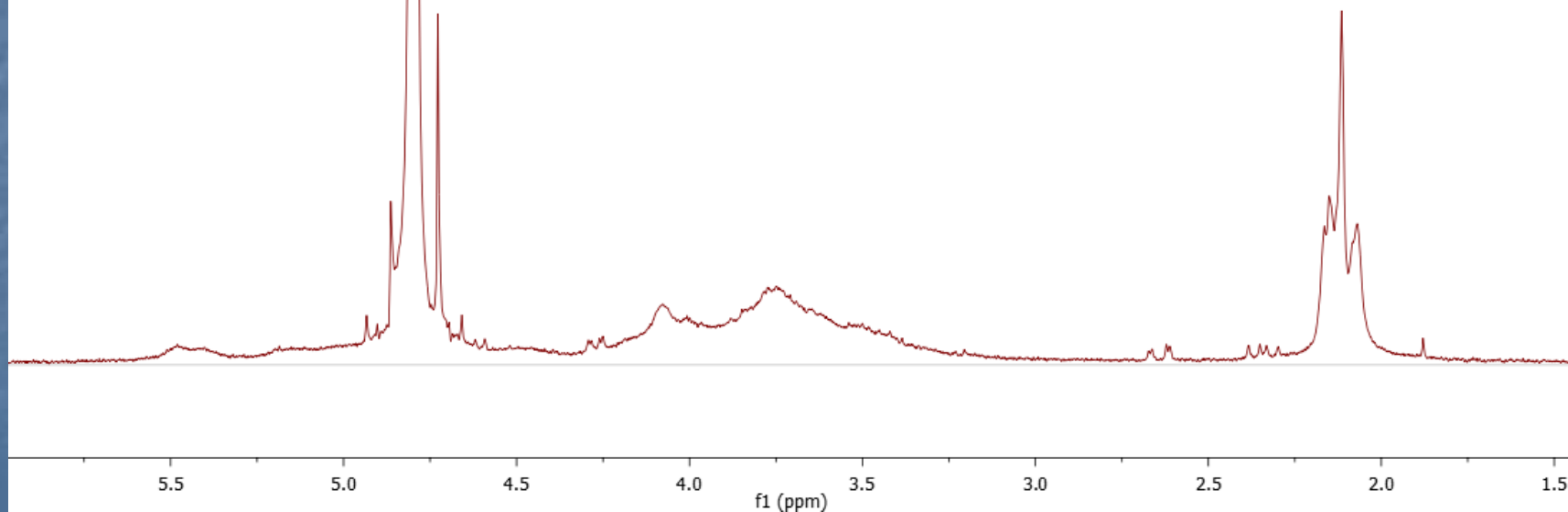


6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5  
f1 (ppm)

Acetylated Polysaccharide 90%+



Scheme 1: Proposed structure for the major component of *Aloe vera* polysaccharide

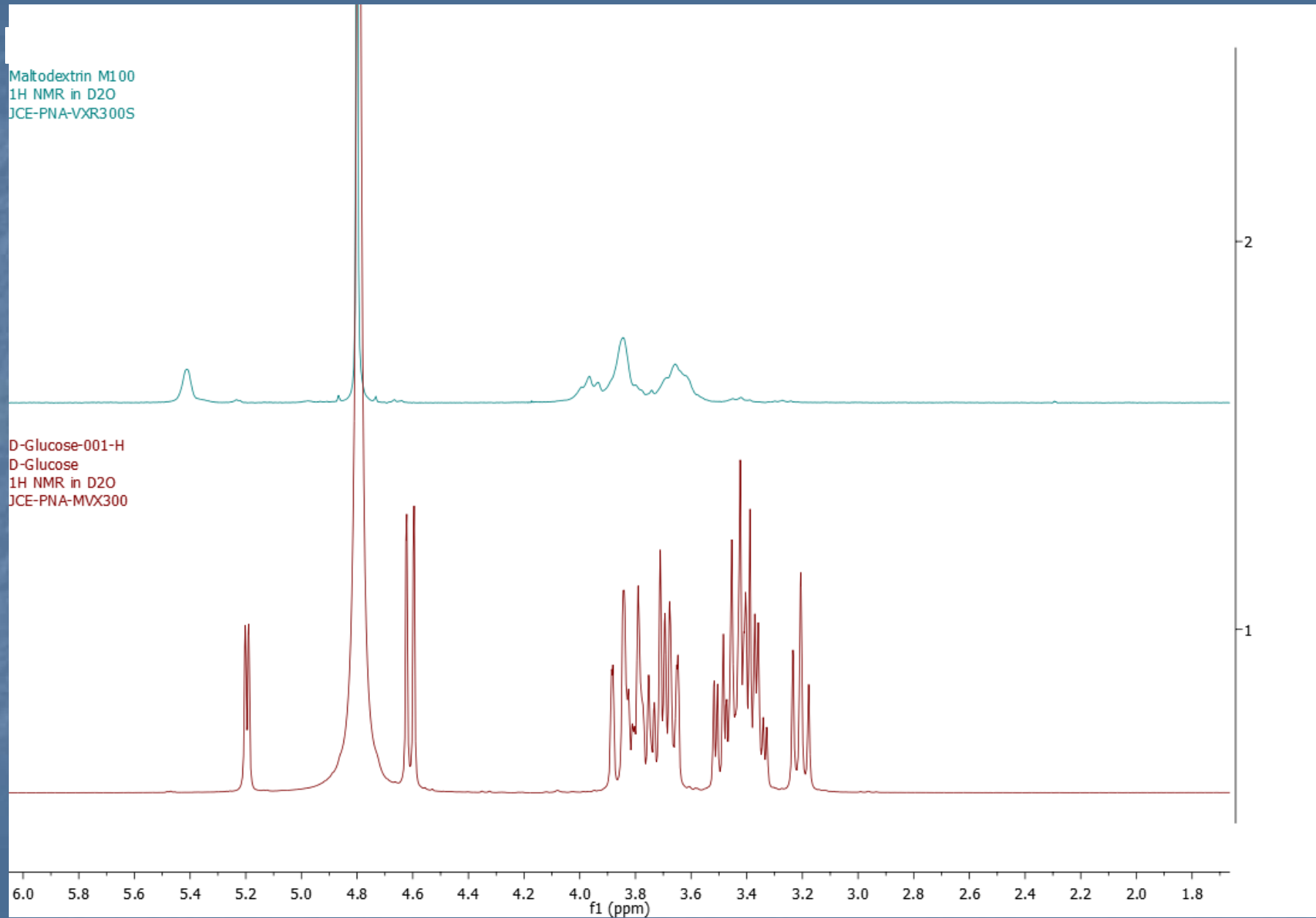


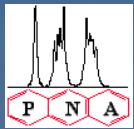
# Chemical Characterization of the Immunomodulating Polysaccharide of *Aloe Vera L.*, W.J.Goux, et al,

<http://www.iasc.org/AloeStructure080604.pdf>

Maltodextrin M100  
1H NMR in D2O  
JCE-PNA-VXR300S

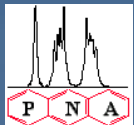
D-Glucose-001-H  
D-Glucose  
1H NMR in D2O  
JCE-PNA-MVX300





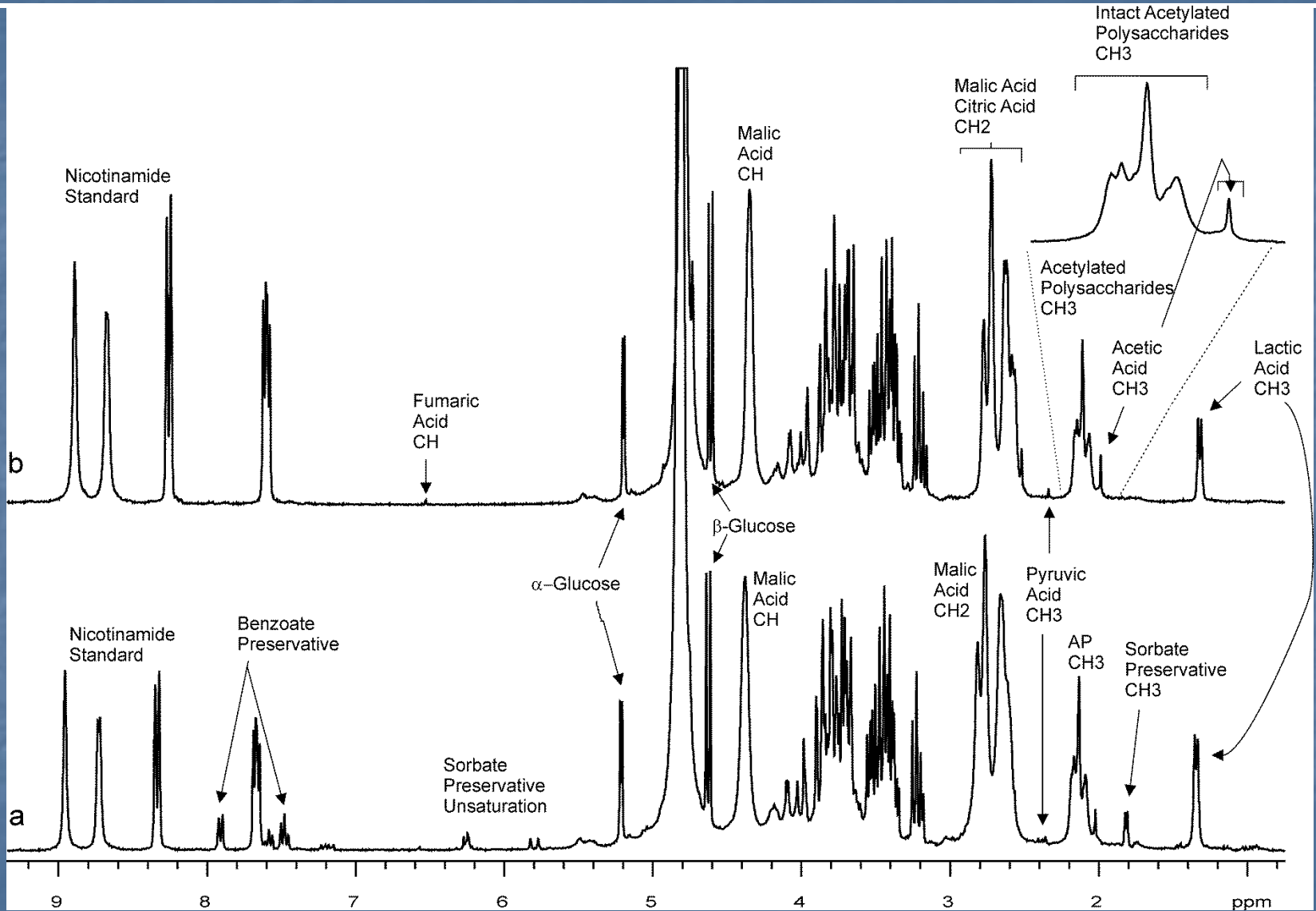
## Aloe Vera Inner Leaf Juice Constituents and Additives that need to be Analyzed and Reported for IASC Certification

<b>Compound</b>	<b>IASC Certification requirement</b>
<b>Acemannan</b>	<b><math>\geq 5\%</math> dry weight</b>
<b>Glucose</b>	<b>Present</b>
<b>Aloin</b>	<b>10 ppm or less in 0.5% aloe vera solids solution, analysed by HPLC or other fit for purpose methodology approved by IASC</b>
<b>Isocitrate</b>	<b><math>\leq 5\%</math> dry weight</b>
<b>Maltodextrin</b>	<b>Must be listed on label and analysis must meet label claims. If undeclared, is considered an adulterant.</b>
<b>Solids</b>	<b><math>\geq 0.46\%</math> in single-strength juice (for example, a 10x concentrate should have <math>\geq 4.6\%</math>)</b>
<b>Ash</b>	<b><math>\leq 40\%</math></b>

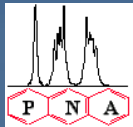


**Characteristic chemical shift values, peak multiplicity, protonated carbon type and N values used for detection and quantitation of the major natural components of aloe vera leaf juice**

<b>Substance</b>	<b>Signal Type and N Parameter</b>	<b>Chemical shift, ppm</b>
<b>Acetylated Polysaccharides</b>	<b>Broad Group of CH<sub>3</sub> Singlets (N=3)</b>	<b>2.0-2.3</b>
<b>Isocitric acid</b>	<b>CH, Doublet (N=1)</b>	<b>4.25</b>
<b>Malic acid</b>	<b>CH, 4 peak multiplet (N=1)</b>	<b>4.45</b>
<b>α-Glucose</b>	<b>CH Doublet (N=1)</b>	<b>4.6</b>
<b>β-Glucose</b>	<b>CH Doublet (N=1)</b>	<b>5.2</b>
<b>Isocitric lactone</b>	<b>CH Doublet (N=1)</b>	<b>5.05</b>

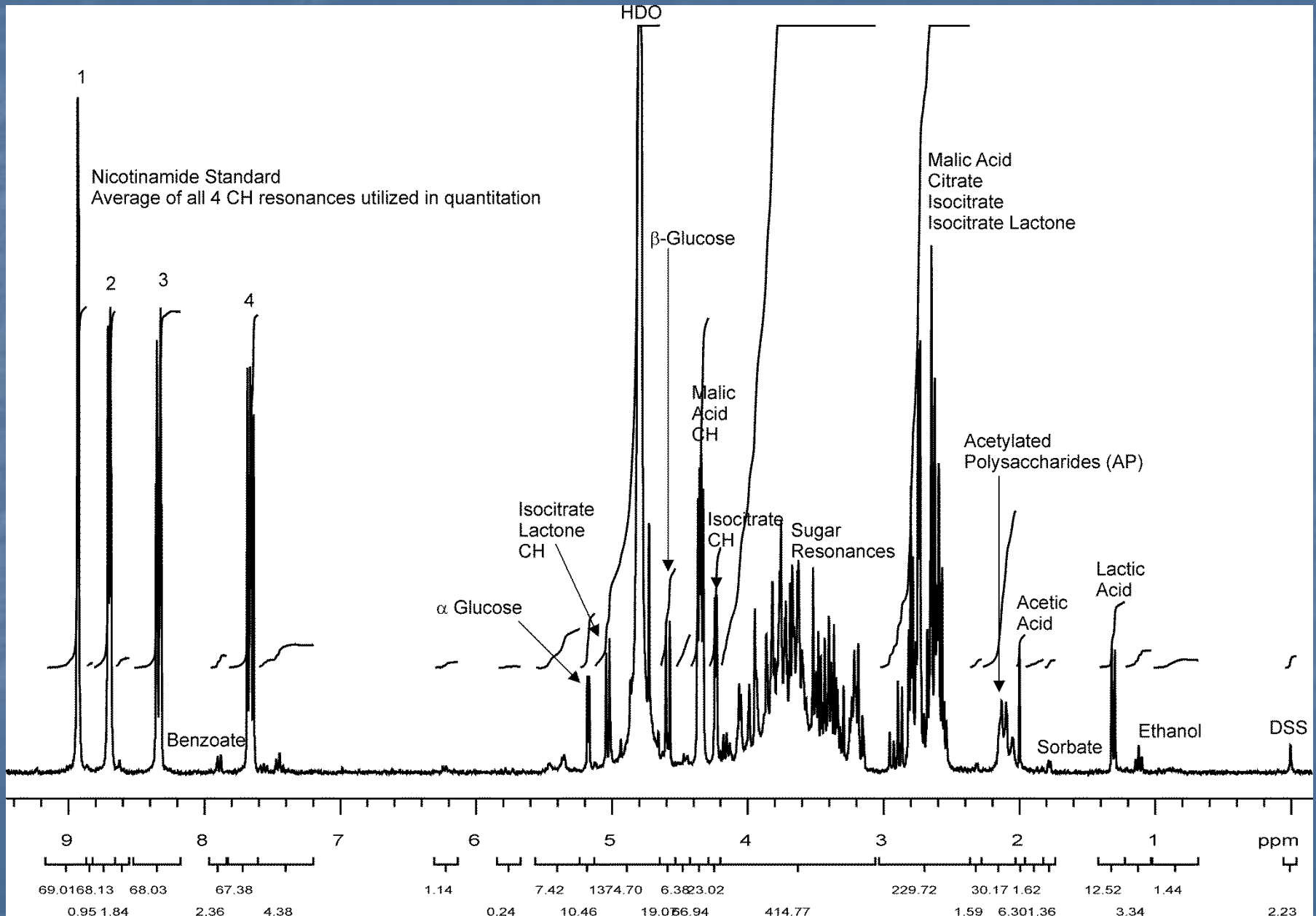


Freeze-Dried Inner Leaf Juices with (a) and without (b) preservatives



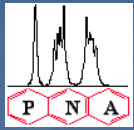
# Commercial freeze-dried 200x aloe vera leaf juice powder

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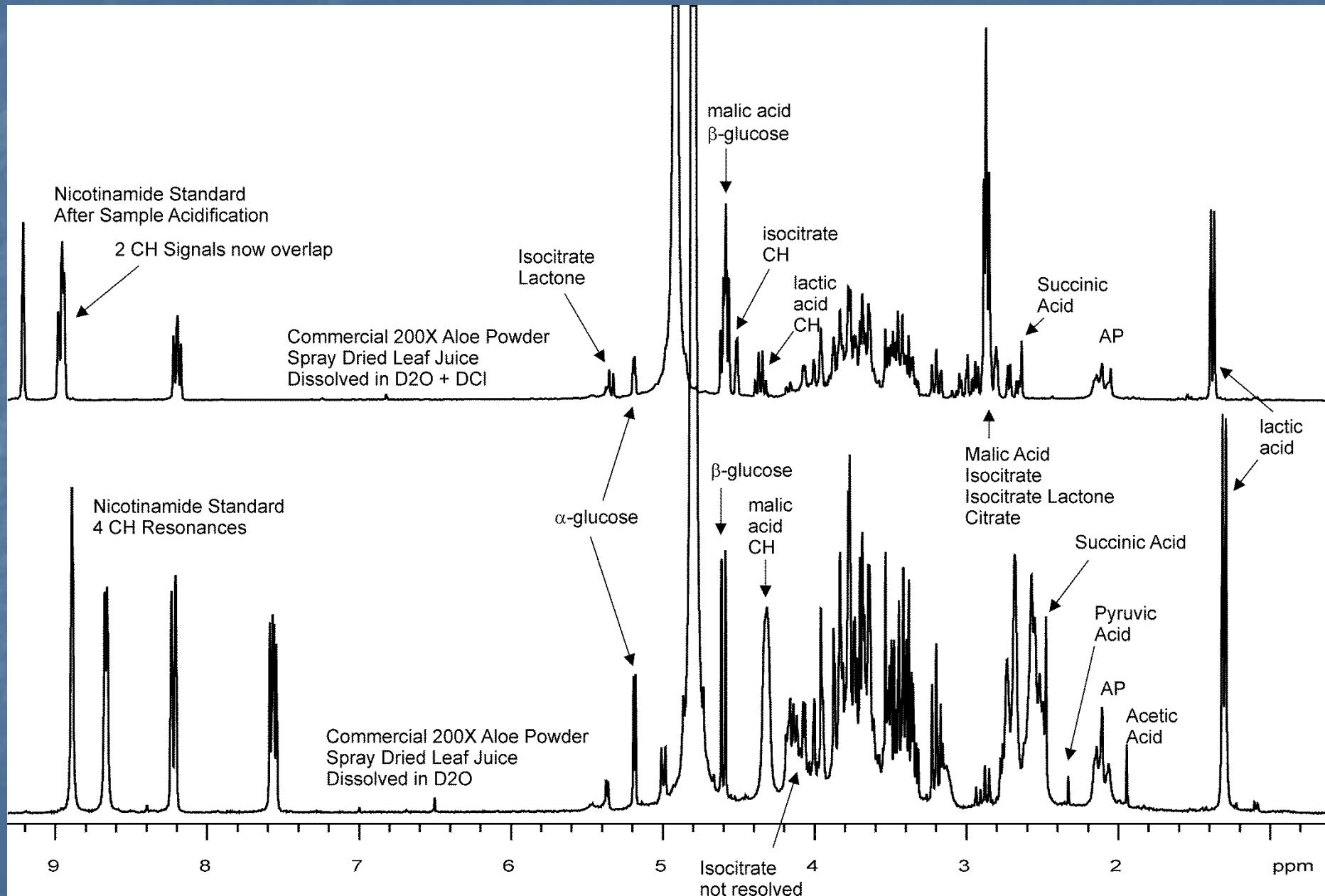
**Chemical shift values, peak and chemistry descriptions, molar conversion factors that can be used for detection and quantitation of aloe vera leaf juice preservatives, additives, and degradation products**

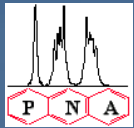
<b>Compound</b>	<b>Type of compound</b>	<b>Signal type</b>	<b>Chemical shift, ppm</b>
Propylene glycol	Additive	CH <sub>3</sub> , doublet (N=3)	1.1
Ethanol	Degradation product or additive	CH <sub>3</sub> , triplet (N=3)	1.15
Lactic acid	Degradation product	CH <sub>3</sub> , doublet (N=3)	1.33
Potassium sorbate	Preservative	CH <sub>3</sub> , doublet (N=3)	1.82
Acetic acid	Degradation product	CH <sub>3</sub> , singlet (N=3)	1.96
Pyruvic acid	Degradation product	CH <sub>3</sub> , singlet (N=3)	2.35
Citric acid	Naturally present or added as pH regulator or preservative	2 x CH <sub>2</sub> , Multiplet (N=4)	2.5-3.0
Succinic acid	Degradation product	2 x CH <sub>2</sub> , singlet (N=4)	2.6
Glycerol	Additive	CH <sub>2</sub> and CH, multiplet	3.5
Glycine	Additive	CH <sub>2</sub> , singlet (N=2)	3.51
Sucrose	Additive	CH, doublet (N=1)	5.4
Fumaric acid	Degradation product	2 x CH, singlet (N=2)	6.5
Sodium benzoate	Preservative	2 x CH, doublet (N=2)	7.95
Formic acid	Degradation product	CH, singlet (N=1)	8.2-8.3



# Freeze Dried Aloe Vera Leaf Juice – Acidification Required for Isocitrate Quantification.

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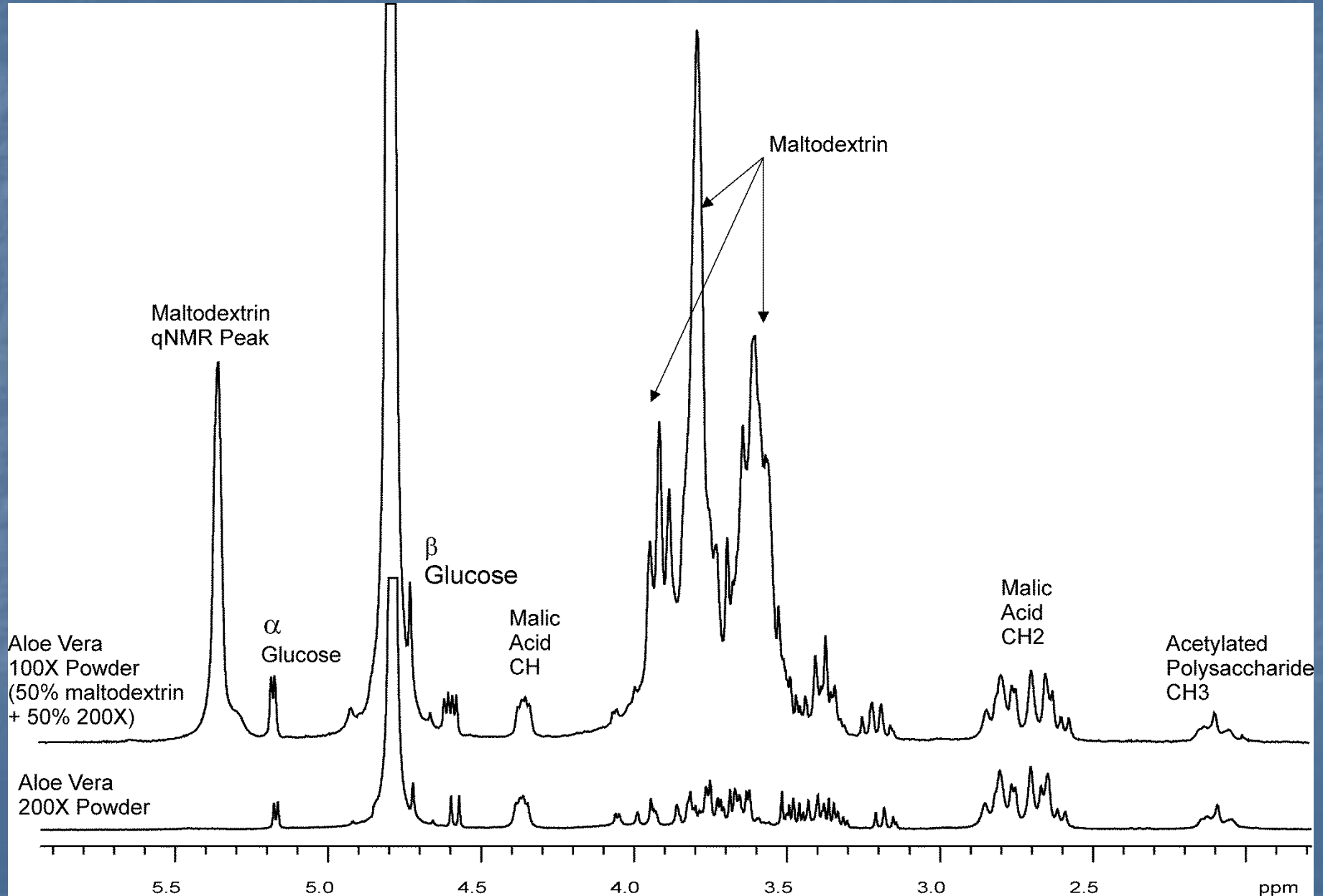


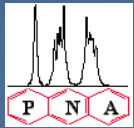


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Commercial 200x inner leaf juice powder

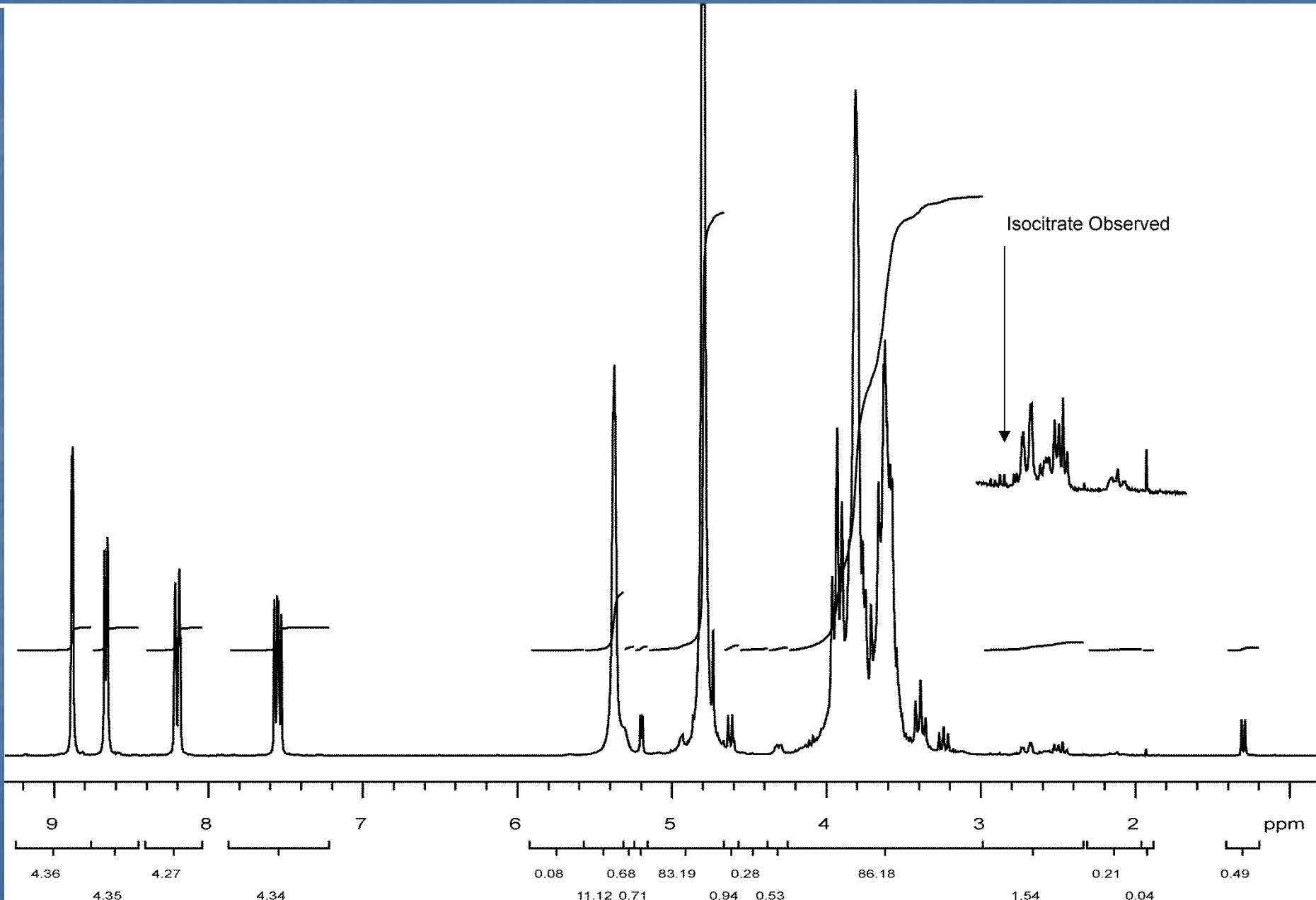
Commercial 100x inner leaf juice powder containing 50% maltodextrin

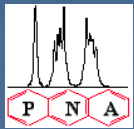




# Aloe Vera Leaf Juice Freeze Dried and Mixed 80% with Maltodextrin

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# Acetylated Polysaccharide Quantitation

Demonstrated that the acetylation of the mannose monomer units is at 78% (Ref. 2, Manna et al., 1993).

Mannose represents 84% of the polysaccharide backbone with the remainder being composed of glucose, galactose, and a few other saccharides (Ref 3., Chow et al., 2005).

The acetylation content and the presence of 16% other saccharides must be taken into account so as not to underestimate the AP content.

The molecular weight of AP is calculated under the assumption that 1 water molecule is removed upon condensation of acetyl with mannosyl monomer, and each mannosyl unit shares in the loss of a single water molecule upon condensation from the predominantly mannose based polysaccharide.

Thus,  $MW_{\text{mannosyl}} = MW_{\text{mannose}} - (MW_{\text{water}})/2 = 180.2 - 18 = 162.2 \text{ g/mol}$

$MW_{\text{AcMann}} = MW_{\text{acetyl group}} + MW_{\text{mannosyl}} - MW_{\text{water}} = 60 + 162.2 - 18 = 204.2 \text{ g/mol}$

Now taking into account the 0.78/1 acetyl/mannosyl residue ratio as well as the presence of ~16% non-mannosyl saccharides in the AP we can calculate the concentration of AP ( $C_{\text{AP}}$ ) by the following equation.

$$C_{\text{AP}} = \left( \frac{W_{\text{Nic}} * I_{\text{AcMann}} * N_{\text{Nic}} * MW_{\text{AcMann}}}{I_{\text{Nic}} * N_{\text{AcMann}} * MW_{\text{Nic}}} + \frac{W_{\text{Nic}} * I_{\text{AcMann}} * N_{\text{Nic}} * MW_{\text{Mann}} * (0.22/0.78)}{I_{\text{Nic}} * N_{\text{AcMann}} * MW_{\text{Nic}}} + \frac{W_{\text{Nic}} * I_{\text{AcMann}} * N_{\text{Nic}} * MW_{\text{Glu}} * (0.16/0.84)}{I_{\text{Nic}} * N_{\text{AcMann}} * MW_{\text{Nic}}} \right) * \frac{1}{W_{\text{sample}}} * 100\%$$

$$= \frac{W_{\text{Nic}} * I_{\text{AcMann}} * N_{\text{Nic}}}{I_{\text{Nic}} * N_{\text{AcMann}} * MW_{\text{Nic}}} * (MW_{\text{AcMann}} + MW_{\text{Mann}} * 0.28 + MW_{\text{Glu}} * 0.19) * \frac{1}{W_{\text{sample}}} * 100\%$$

$$= \frac{W_{\text{Nic}} * I_{\text{AcMann}} * 4}{I_{\text{Nic}} * W_{\text{sample}} * 3 * 122.1} * (204.2 + (162.2 * 0.47)) * 100\%$$

$$= \frac{W_{\text{Nic}} * I_{\text{AcMann}}}{I_{\text{Nic}} * W_{\text{sample}}} * 3.06 * 100\%$$

$AcMann$  = acetylmannosyl,  $Mann$ =Mannosyl,  $Glu$ =glucosyl

$N$  = number of protons in the group - molar conversion factor -  $N_{\text{Nic}}=4$ ,  $N_{\text{AP}}=3$

$C_{\text{AP}}$  = content of acetylated polysaccharides (AP) in the sample, wt%

$W_{\text{Nic}}$  = weight of added internal standard (mg),  $W_{\text{sample}}$  = weight of sample (mg)

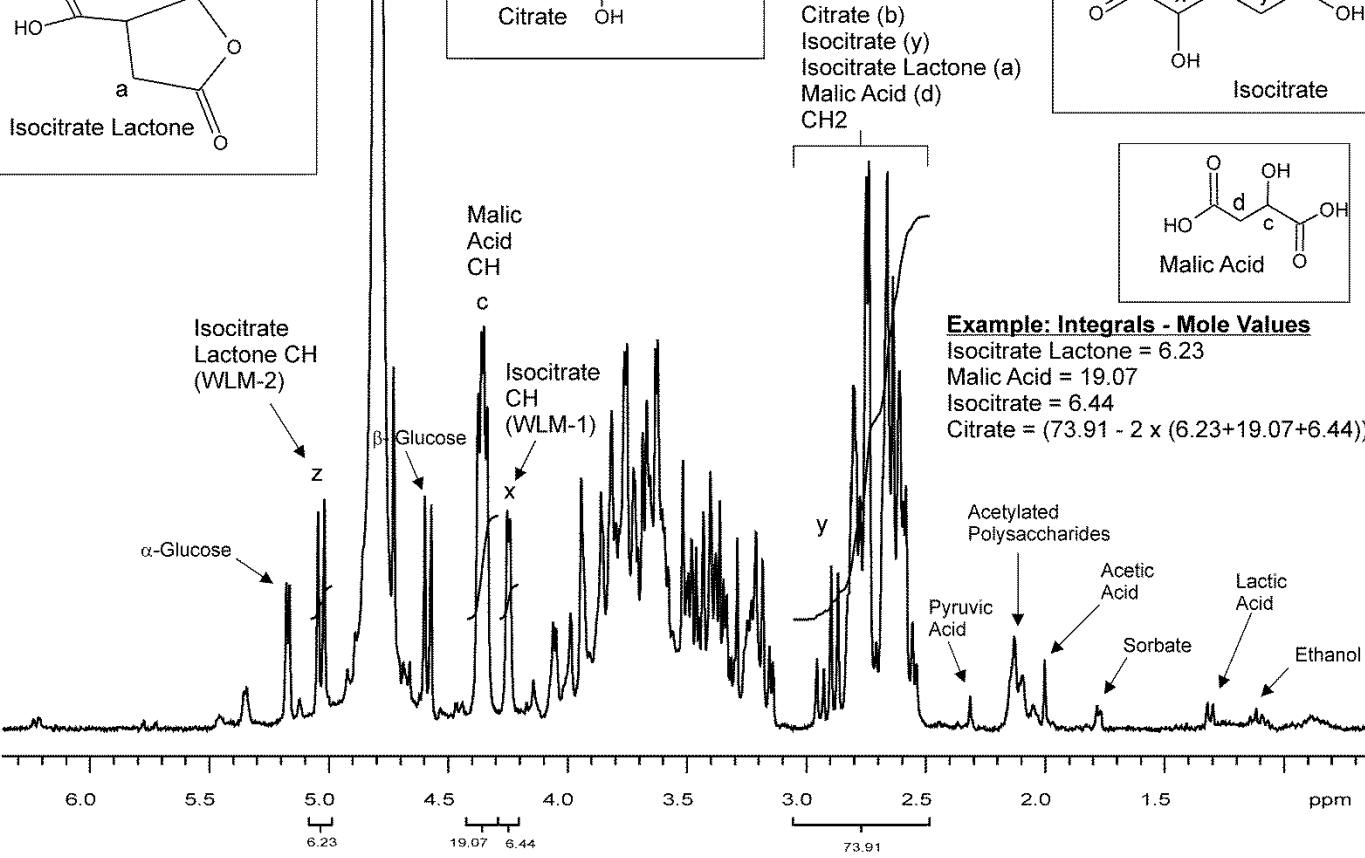
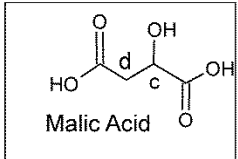
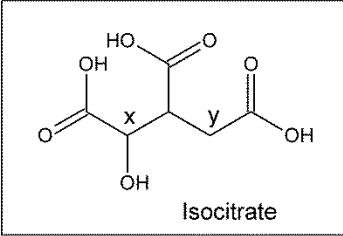
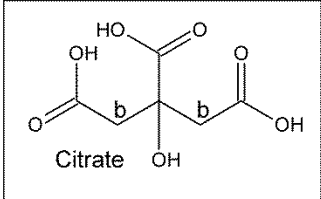
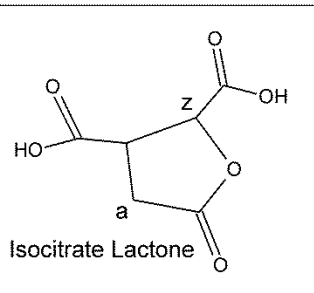
$I_{\text{AcMann}}$  = integration area of acetylation methyls (multiple peaks in 2.0-2.3 ppm region)

$I_{\text{Nic}}$  = sum of integration areas of the 4 aromatic CH peaks of the nicotinamide standard

$MW_{\text{AcMann}}$  = Molecular Weight of acetylated polysaccharides (204.2 g/mol)

$MW_{\text{Nic}}$  = molecular weight of nicotinamide standard (122.1 g/mol)

$MW_{\text{Mann}}=MW_{\text{Glu}}=162.2$



**Example: Integrals - Mole Values**  
 Isocitrate Lactone = 6.23  
 Malic Acid = 19.07  
 Isocitrate = 6.44  
 Citrate = (73.91 - 2 x (6.23+19.07+6.44))/4 = 2.61

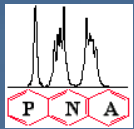
# Isocitrate Quantitation

1H NMR of Freeze Dried Aloe Vera Leaf Juice: Assignments and Explanation of quantification of Isocitrate (WLM-1), Isocitrate Lactone (WLM-2), Citrate, and Malic Acid.  
 Note: peaks marked z, x, and y do not appear in aloe vera inner leaf materials.

$$C_{ICA} \text{ (wt\%)} = 100\% * (W_{Nic} * I_{ICA} * N_{Nic} * MW_{ICA}) / (I_{Nic} * N_{ICA} * MW_{Nic} * W_{Sample})$$

$$= 100\% * (W_{Nic} * I_{ICA} * 4 * 192.1) / (I_{Nic} * 1 * 122.1 * W_{Sample}) = 100\% * (W_{Nic} * I_{ICA} / I_{Nic} * W_{Sample}) * 6.29$$

- $C_{ICA}$  = content of isocitrate in the sample, weight%
- $W_{Nic}$  = weight of added nicotinamide internal standard (mg),  $W_{Sample}$  = weight of sample (mg)
- $I_{ICA}$  = integration area of CH proton peak of isocitrate (doublet at 4.45 ppm (dissolved in D<sub>2</sub>O and acidified with DCI))
- $I_{Nic}$  = sum of integration areas of the 4 aromatic CH peaks of the nicotinamide standard
- $MW_{ICA}$  = Molecular Weight of Isocitrate (192.1 g/mol),  $MW_{Nic}$  = molecular weight of nicotinamide standard (122.1 g/mol)
- $N$  = number of protons in the group - molar conversion factor,  $N_{Nic}=4$ ,  $N_{ICA}=1$



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**Quantitative Proton Nuclear Magnetic Resonance Spectrometry ( $^1\text{H}$  NMR)  
for Determination of Acetylated Polysaccharides, Glucose, Maltodextrin,  
and Isocitrate in Aloe Vera Leaf Juice**

**Included in the American Herbal Pharmacopoeia Monograph on Aloe Vera  
Leaf, Leaf Juice & Inner Leaf Juice**